CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the October/November 2015 series

0653 COMBINED SCIENCE

0653/23

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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Page	2	Mark Scheme	Syllabus	Paper
		Cambridge IGCSE – October/November 2015	0653	23
1 (a)	tr	ronchus correctly labelled ; achea correctly labelled ; arynx correctly labelled ;		[3]
(b)		maller airway diameter ; resence of mucus obstructing flow ;		[2]
(c)	(i	(i) $(12.5 - 5.8) = 6.7 \text{ (dm}^3/\text{min)}$;		
		$(\frac{6.7}{5.8} \times 100) = 115 \text{ or } 116(\%);$		[2]
	(ii	i) breathe more quickly ; take deeper/bigger breaths ;		[2]
		take deeper, sigger steame,		[Total: 9]
2 (a)	d	ecreases and endothermic ;		[1]
(b)	h	ydrogen ;		[1]
(c)		ate of reaction increases/bubbles produced more rapidly/greater temphange/increases rate of temperature change/reaction will finish faste		[max 1]
(d)	С	opper below hydrogen in the reactivity series ; opper will not displace hydrogen from acid/will not react ; o temperature change ;		
		opper is less reactive than magnesium ;		[max 2]
				[Total: 5]
3 (a)	(i	i) weight/gravitational force ;		[1]
	(ii	i) newton;		[1]
(b)	-	otential (energy) → kinetic (energy) ; inetic (energy) → potential (energy) ;		[2]
(c)		a to B accelerating ; to C constant speed ;		[2]
(d)		distance =) speed \times time 9 \times 2 ; 18 (m) ;		[2]
				[Total: 8]

Mark Scheme

Syllabus

Paper

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Page 3	Mark Scheme	Syllabus	Paper
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- 4 (a) (i) contains the correct proportions of nutrients for an individual; [1]
 - (ii) to prevent scurvy/AVP; [1]
 - (iii) (vitamin C) is a small molecule/only large molecules need to be digested;can be absorbed without being broken down;(vitamin C) must be soluble;[max 2]
 - (b) (i) orange and kiwifruit;

(because) **both** have average portion values greater than 60 mg;

OR

orange **AND** has average portion value 70 g/greater than 60 mg;

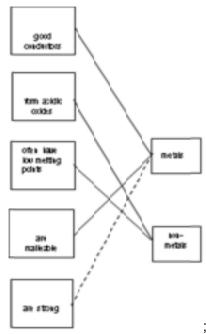
kiwifruit **AND** has average portion value 74 g/greater than 60 mg; [max 2]

(ii) $\frac{60}{28}$; $\times 100 = 214.3 (g)$; [2]

- (c) (i) acid would attack/destroy the enamel; [1]
 - (ii) regularly clean teeth/use mouth wash; [1]

[Total: 10]

5 (a)



all 4 correct for 2 marks, 3 or 2 correct for 1 mark

[2]

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observation explanation
(bubbles of gas) hydrogen;
(indicator changes from green to purple) alkaline solution;

[2]

(d) (i)

substance	diagram
(chlorine)	Α
(sodium)	В
(sodium chloride)	С

all 3 correct for 2 marks, 2 or 1 correct for 1 mark

[2]

(ii) sodium atoms lose electrons; chlorine atoms gain electrons; electrons are transferred from sodium to chlorine;;

[max 2]

(iii) opposite charges (attract);

[1]

[2]

(iv) (ZnCl₂) symbols; numbers;

[Total: 12]

6 (a) (i) (speed =)
$$\frac{30}{10}$$
 = 3 (m/s);

[1]

(ii) (frequency =
$$\frac{10}{20}$$
) = 0.5;

Hz/hertz;

[2]

(b) (distance moved =
$$2 \times \text{amplitude}$$
) = 1.0 (m);

[1]

[1]

[max 1]

[Total: 6]

P	age	5	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2015	0653	23
7	(a)	(i)	reduces soil erosion ; roots of trees stabilise soil ;		[2]
		(ii)	provides habitats; provides shelter/protection from sun/rain/predators; maintains food supply;		[max 2]
	(b)	(i)	bacteria/microbes use it ; for respiration ;		[2]
		(ii)	(sewage) blocks/reduces light; light is needed for photosynthesis;		[2]
					[Total: 8]
8	(a)	(i)	copper carbonate + hydrochloric acid → copper chloride + carbon dioxide + water water in product row ;		
			rest of equation correct;		[2]
		(ii)	limewater; milky/cloudy;		[2]
	(b)	(i)	(cathode) copper; chlorine;		[2]
		(ii)	an ionic compound ; an element ; a covalent compound ;		[3]
					[Total: 9]
9	(a)	(the	ermal) expansion ;		[1]
	(b)	(i)	conduction;		[1]
		(ii)	convection;		[1]
	(c)	(i)	evaporation;		[1]
		(ii)	molecules gain more energy on heating; molecules with enough/high/higher energy escape;		[2]
					[Total: 6]

Mark Scheme

Syllabus

Paper

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			Cambridge IGCSE – October/November 2015	0653	23
10	(a)	(i)	(charging by) friction/rubbing with cloth/AVP;		[1]
	 (b) plastic is an insulator; so charge wouldn't leak away; OR metal is conductor; so charge would leak away/earthed by student; 			[2]	
	(c)	(i)	flow of charge ;		[1]
		(ii)	workable circuit with additional heater in series/parallel and at leas	t one switch	1

Mark Scheme

[Total: 7]

[3]

Syllabus

Paper

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and no short circuit; two heaters in parallel;

both switches on parallel branches;